

## The Equation Lab

1. a. iron + oxygen  $\rightarrow$  iron(III) oxide

b. methane (CH<sub>4</sub>) + oxygen  $\rightarrow$  carbon dioxide + water

2. Magnesium + oxygen  $\rightarrow$  magnesium oxide

3. Copper + oxygen  $\rightarrow$  copper (II) oxide

4. a. Zinc + Hydrogen chloride  $\rightarrow$  zinc chloride + hydrogen

b. hydrogen + oxygen  $\rightarrow$  water

c. Cellulose(C<sub>6</sub>H<sub>10</sub>O<sub>5</sub>) + oxygen  $\rightarrow$  carbon dioxide + water

5. sodium hydrogen carbonate + HCl  $\rightarrow$  sodium chloride + hydrogen carbonate

6. hydrogen peroxide (H<sub>2</sub>O<sub>2</sub>)  $\rightarrow$  water + oxygen

7.  $C_3H_7OH + \text{oxygen} \rightarrow \text{carbon dioxide} + \text{water}$

8.  $\text{potassium iodide} + \text{lead (II) nitrate} \rightarrow \text{lead iodide} + \text{potassium nitrate}$

9.  $\text{Sodium hydrogen carbonate} \rightarrow \text{sodium oxide} + \text{water} + \text{carbon dioxide}$

10.  $\text{water} + \text{carbon dioxide} \rightarrow \text{hydrogen carbonate}$

11.  $\text{Aluminum} + \text{copper(II) chloride} \rightarrow \text{aluminum chloride} + \text{copper}$

12.  $\text{Magnesium} + \text{hydrogen nitrate} \rightarrow \text{magnesium nitrate} + \text{hydrogen}$

13.  $\text{hydrogen phosphate} + \text{sodium carbonate} \rightarrow \text{sodium phosphate} + \text{hydrogen carbonate}$